

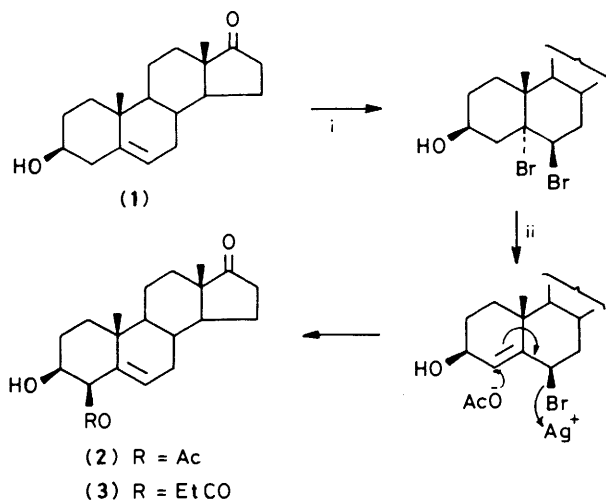
Neighbouring Group Participation in the Allylic Oxidation of a Δ^5 -Steroid

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When 3 β -acetoxy- Δ^5 -steroids are treated with bromine and silver acetate, the product 3 β -hydroxy-4 β -acetate is formed by intramolecular acetoxy group migration via a 3 β ,4 β -acetoxylium ion.

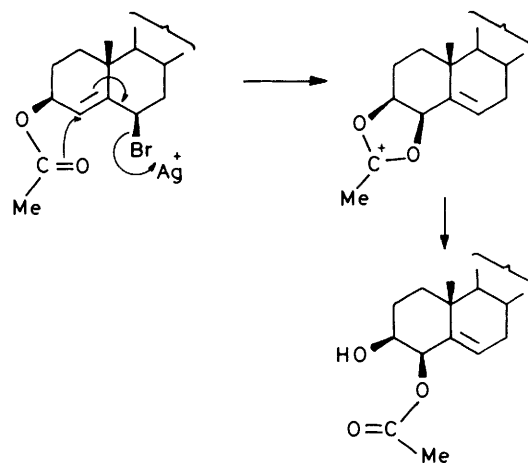
The allylic acetoxylation of Δ^5 -steroids [e.g. (1)] by reaction with bromine in chloroform followed by treatment with silver acetate in pyridine at -60°C provides a simple route to some 4 β -acetoxy- Δ^5 -steroids [e.g. (2)].¹ The reaction (see Scheme 1) is



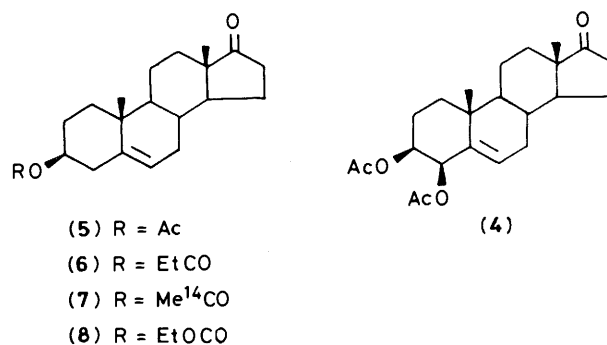
Scheme 1. Reagents: i, Br₂; ii, AgOAc, pyridine

limited to a certain extent by the nature of the 3 β -substituent.⁴ In the course of our studies on the scope and limitations of this reaction we observed that 3 β -acetoxyandrost-5-en-17-one (5) gave the 4 β -acetoxy-3 β -hydroxy- Δ^5 -steroid (2) rather than the 3 β ,4 β -diacetate (4). The ¹H n.m.r. spectrum of this compound showed the 3 α -H resonance as a multiplet (δ 4.65; $w_{\frac{1}{2}}$ 20 Hz) and the 4 α -H resonance as a lower field doublet (δ 5.4; J 4 Hz). This contrasts with the isomeric 3 β -acetoxy-4 β -hydroxyandrost-5-en-17-one in which the 3 α -H resonance was a multiplet (δ 4.65; $w_{\frac{1}{2}}$ 14 Hz) whilst the 4 α -H resonance appeared at δ 4.2 as a doublet (J 4 Hz). In the 3 β ,4 β -diacetate these signals appear at δ 4.73 and 5.53 respectively.³ An interpretation of this result (see Scheme 2) is that the 3 β -acetate is participating in the *syn* allylic displacement of the 6 β -bromide intermediate in the reaction with the formation of a 3 β ,4 β -acetoxylium ion which then undergoes hydrolysis. In this paper we present some evidence to support this reaction scheme.⁴

The reaction (see Scheme 1) proceeds by the addition of bromine followed by the stepwise elimination and allylic substitution of the 5 α - and 6 β -bromine atoms. Treatment of both 3 β -acetoxy-5 α ,6 β -dibromoandrost-17-one (9) and 3 β -acetoxy-6 β -bromoandrost-4-en-17-one (11) (obtained by dehydration *in situ* of the corresponding 5 α -hydroxy-6 β -bromo steroid with thionyl chloride) with silver acetate in pyridine gave the known 4 β -acetoxy-3 β -hydroxyandrost-5-en-17-one (2).¹ The rearrangement of the 3 β -ester was independent of the anion since treatment of the 3 β -propionate (6) with bromine followed by silver acetate gave the 3 β -hydroxy-4 β -propionate (3) rather



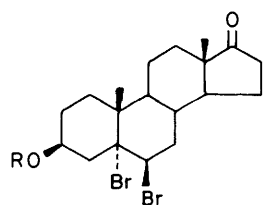
Scheme 2.



than the 4 β -acetate (2). The authentic 4 β -propionate (3) was obtained from the reaction of the 5 α ,6 β -dibromide (10) of the 3 β -alcohol, dehydroisoandrosterone, with silver propionate. Indeed, when silver oxide was used in place of the silver acetate, the 3 β -acetate (5) still gave the 4 β -acetoxy-3 β -hydroxy steroid (2) although in lower yield.

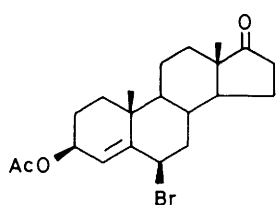
The confirmation that an intramolecular acyl migration to C-4 had occurred was obtained when [¹⁴C]-3-acetoxyandrost-5-en-17-one (7) (specific activity, 11.4 kBq mmol⁻¹) was used as the substrate. The resultant 4 β -acetoxy-3 β -hydroxyandrost-5-en-17-one (2) had a specific activity of 10.3 kBq mmol⁻¹ having retained 89% of the radioactivity of the initial acetate (7). The intermediate acyloxonium ion was trapped by using 3 β -ethoxy-carbonylandrost-5-en-17-one (8) as a substrate. The product from this reaction was the cyclic 3 β ,4 β -carbonate (12) which was identified by its i.r. (ν_{max} 1780 cm⁻¹) and ¹³C n.m.r. (δ_{C} 155.0, O₂C=O) spectra.

The overall stereochemistry of the reaction followed from the fate of a 4 β -deuterium label. [4 β ,17 α -²H₂]-3 β ,17 β -Diacetoxyandrost-5-ene (13) was prepared by an adaption of a known

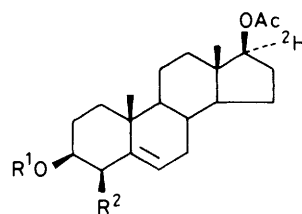
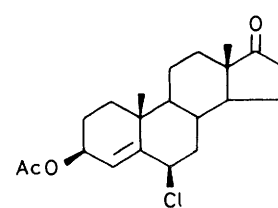


(9) R = Ac

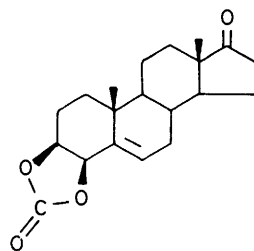
(10) R = H



(11)

(13) R¹ = Ac, R² = ²H

(14)



(12)

(15) R¹ = H, R² = OAc

procedure^{5,6} involving the reduction of 3 β -acetoxy-6 β -chloroandrost-4-en-17-one (14) with lithium aluminium deuteride and subsequent acetylation of the product. The ²H n.m.r. spectrum of the diacetate showed signals at δ (²H) 2.29 (4 β -²H) and 4.59 (17 α -²H). The deuterium-labelled steroid was treated with bromine and subsequently silver acetate in pyridine. The ²H n.m.r. spectrum of the resultant 4 β ,17 β -diacetoxy-3 β -hydroxyandrost-5-ene (15) showed only one signal at δ 4.58 which was attributed to the 17 α -²H. There was no signal at δ 5.37 which could be assigned to the 4 α -²H. This 4 α -H signal remained in the ¹H n.m.r. spectrum of the product. Hence the 4 β -acetate has replaced a 4 β -deuterium label and the reaction has proceeded with an overall retention of configuration.

The reaction sequence can therefore be formulated as the addition of bromine to the Δ^5 -double bond, the *trans* elimination of the 4 β -proton and the 5 α -bromine atom, followed by a *syn* intramolecular silver ion assisted substitution of the 6 β -bromine atom by the 3 β -acetoxy group to form a 3 β ,4 β -acetoxylium ion. This then collapses to afford the 4 β -acetoxy-3 β -hydroxy steroid. The intervention of a 3 β ,4 β -acetoxylium ion has been invoked in a number of previous studies including the allylic displacement of a Δ^4 -6-chloride,⁵ the opening of a 3 β -acetoxy-4 α ,5 α -epoxide,⁷ and the addition of hypobromous acid to a 3 β ,17 β -diacetoxyoestr-4-ene.⁸ An interesting feature of each of these rearrangements is that the intermediate 3 β ,4 β -acetoxylium ion opens to give the less stable 4 β -axial acetate, a feature which has recently been rationalized in terms of stereoelectronic effects.⁹

Experimental

Acetoxylation of 3 β -Acetoxyandrost-5-en-17-one (5).—3 β -Acetoxyandrost-5-en-17-one (4 g) in chloroform (30 ml) was cooled to -60°C . Bromine (0.7 ml) was added with swirling followed by silver acetate (6 g) in pyridine (9 ml). The reaction mixture was stirred and allowed to attain room temperature in the dark during 24 h. The mixture was poured into dilute hydrochloric acid and the steroids were extracted into chloroform. The extract was washed with water and aqueous sodium hydrogen carbonate and dried. The solvent was evaporated and the residual gum was chromatographed on silica. Elution with 25% ethyl acetate–light petroleum gave 4 β -acetoxy-3 β -hydroxyandrost-5-en-17-one (2) (2.5 g) which crystallized from acetone as needles, m.p. $198.5\text{--}200^\circ\text{C}$, $[\alpha]_{\text{D}} -62^\circ$ (c 0.3 in CHCl_3),¹ m.p. $192\text{--}193^\circ\text{C}$, $[\alpha]_{\text{D}} -60.7^\circ$.

The product was identified by its i.r. and n.m.r. spectra. Under similar conditions 3 β -acetoxy-5 α ,6 β -dibromoandrost-17-one (500 mg) gave 4 β -acetoxy-3 β -hydroxyandrost-5-en-17-one (200 mg).

Preparation of 3 β -Acetoxy-6 β -bromo-5 α -hydroxyandrost-17-one.—A solution of 3 β -acetoxy-5 α ,6 α -epoxyandrost-17-one (1 g) in chloroform (10 ml) was treated with an ice-cold solution of lithium bromide (1 g) in water (1 ml) and 48% hydrobromic acid (1 ml) for 12 h. Chloroform was added and the organic layer was separated, and washed with water and aqueous sodium hydrogen carbonate. The solvent was evaporated to afford 3 β -acetoxy-6 β -bromo-5 α -hydroxyandrost-17-one (1.1 g) which crystallized from acetone as small prisms, m.p. $180\text{--}183^\circ\text{C}$, $[\alpha]_{\text{D}} -14.3^\circ$ (Found: C, 60.5; H, 7.5. $\text{C}_{21}\text{H}_{31}\text{BrO}_4$ requires C, 59.0; H, 7.3%); ν_{max} . 3 540, 1 740, and 1 730 cm^{-1} ; δ 0.90 (3 H, s, 18-H), 1.36 (3 H, s, 19-H), 2.03 (3 H, s, OAc), 4.0 (1 H, m, $w_{\frac{1}{2}}$ 4 Hz, 6-H), and 5.08 (1 H, m, $w_{\frac{1}{2}}$ 30 Hz, 3-H).

Dehydration of 3 β -Acetoxy-6 β -bromo-5 α -hydroxyandrost-17-one and the Reaction with Pyridine and Silver Acetate.—The bromo steroid (1 g) in dry pyridine (15 ml) was treated with thionyl chloride (0.5 ml) at 0°C . The mixture was allowed to attain room temperature and then poured onto ice-water. The steroids were extracted with diethyl ether. The extract was washed thoroughly with aqueous copper sulphate and saturated sodium chloride solution. The extract was dried (Na_2SO_4) and the solvent was evaporated to give a gum which was redissolved in methylene dichloride and cooled to -60°C . Silver acetate (1 g) in pyridine (2 ml) was added and the mixture was stirred at -60°C for 1 h and then allowed to attain room temperature in the dark during 40 h. The mixture was diluted with methylene dichloride and filtered. The solution was washed with dilute hydrochloric acid, water, and aqueous sodium hydrogen carbonate. The solvent was evaporated and the residue was chromatographed on silica. Elution with 50% ethyl acetate–light petroleum gave 4 β -acetoxy-3 β -hydroxyandrost-5-en-17-one (290 mg) which was identified by its i.r. and n.m.r. spectra.³

Reaction of 3 β -Propionyloxyandrost-5-en-17-one with Bromine and Silver Acetate.—3 β -Propionyloxyandrost-5-en-17-one (2 g)¹⁰ in chloroform (10 ml) was treated with bromine (0.4 ml) at -60°C . Silver acetate (3 g) in pyridine (6 ml) was then added and the mixture was allowed to attain room temperature in the dark during 20 h, when poured into dilute hydrochloric acid and the steroid extracted with chloroform. The extract was washed with water and aqueous sodium hydrogen carbonate. The solvent was evaporated and the residue was chromatographed on silica. Elution with 25% ethyl acetate–light petroleum gave 3 β -hydroxy-4 β -propionyloxyandrost-5-en-17-one (700 mg), m.p. $155\text{--}158^\circ\text{C}$, identical (i.r. and n.m.r.) with the material prepared² by the action of silver propionate and bromine on dehydroisoandrosterone.

Reaction of 3 β -Acetoxyandrost-5-en-17-one with Bromine and Silver Oxide.—Bromine (0.9 ml) was added to a solution of 3 β -acetoxyandrost-5-en-17-one (500 mg) in chloroform (5 ml) at -60°C . Moist silver oxide (1 g) in pyridine (1 ml) was added in two equal portions and the mixture was stirred at room temperature in the dark during 6 days. The mixture was poured into dilute hydrochloric acid and the steroids were recovered in ethyl acetate. The extract was washed with water and aqueous sodium hydrogen carbonate. The solvent was evaporated and the residue chromatographed on silica. Elution with 50% ethyl acetate–light petroleum gave 4 β -acetoxy-3 β -hydroxyandrost-5-en-17-one (43 mg) which was identified by its i.r. and n.m.r. spectra.³

Reaction of [1'- ^{14}C]-3 β -Acetoxyandrost-5-en-17-one with Bromine and Silver Acetate.—[1'- ^{14}C]-3 β -Acetoxyandrost-5-en-17-one was prepared from [1'- ^{14}C]acetic anhydride and pyridine. The labelled steroid (1 g, specific activity, 11.4 kBq mmol^{-1}) was treated with bromine (0.2 ml) in chloroform (10 ml) at -60°C . Silver acetate (1 g) in pyridine (2 ml) was added and the mixture was stirred for 0.5 h. A further portion of silver acetate (0.5 g) in pyridine (1 ml) was then added and the solution was allowed to attain room temperature in the dark during 24 h. The products were recovered as above to afford 4 β -[1'- ^{14}C]acetoxy-3 β -hydroxyandrost-5-en-17-one (500 mg) which was crystallized to constant radioactivity. It had specific activity 10.3 kBq mmol^{-1} .

3 β -Ethoxycarbonylandrost-5-en-17-one.—3 β -Hydroxyandrost-5-en-17-one (2 g) in pyridine (15 ml) was treated with ethyl chloroformate (5 ml) overnight. The solution was poured onto ice–water and the steroid was recovered in diethyl ether. The extract was washed thoroughly with aqueous copper sulphate, saturated sodium chloride solution, and dried. The solvent was evaporated to afford 3 β -ethoxycarbonylandrost-5-en-17-one (2.4 g) which crystallized from acetone as prisms, m.p. $179\text{--}182^{\circ}\text{C}$, $[\alpha]_{\text{D}} + 1^{\circ}$ (c 0.2) (Found: C, 73.3; H, 8.9. $\text{C}_{22}\text{H}_{32}\text{O}_4$ requires C, 73.3; H, 8.95%); v_{max} 1 740, 1 665 cm^{-1} ; δ 0.89 (3 H, s, 18-H), 1.04 (3 H, s, 19-H), 1.31 (3 H, t, J 7 Hz, OCH_2CH_3), 4.19 (2 H, q, J 7 Hz, OCH_2CH_3), 4.28 (1 H, m, 3-H), and 5.43 (1 H, m, 6-H).

Reaction of 3 β -Ethoxycarbonylandrost-5-en-17-one with Bromine and Silver Acetate.—Bromine (0.4 ml) was added to a solution of 3 β -ethoxycarbonylandrost-5-en-17-one (2 g) in methylene dichloride (15 ml) at -60°C . Silver acetate (3 g) in pyridine (6 ml) was then added during 10 min and the mixture was allowed to attain room temperature during 3 days. The solution was filtered, diluted with methylene dichloride, washed with saturated copper sulphate solution, and water, dried and evaporated to give a residue which was chromatographed on silica. Elution with 20% ethyl acetate–light petroleum gave the starting material (1.2 g). Further elution with 35% ethyl acetate–light petroleum gave the 3 β ,4 β -cyclic carbonate of 3 β ,4 β -dihydroxyandrost-5-en-17-one (550 mg) which crystallized from ethyl acetate–light petroleum as plates, m.p. $229\text{--}232^{\circ}\text{C}$, and which was identified by its i.r. and n.m.r. spectra.¹¹

Preparation of [4 β ,17 α - $^2\text{H}_2$]-3 β ,17 β -Diacetoxyandrost-5-ene.—3 β -Acetoxy-4 β -hydroxyandrost-5-en-17-one (2 g) (prepared by the partial acetylation of the 3,4-diol)¹ in dry diethyl ether (80 ml) and pyridine (0.4 ml) was treated with thionyl chloride (0.5 ml) in diethyl ether (8 ml). The suspension was stirred for 24 h. The mixture was washed with ice-cold dilute hydrochloric acid, water, and aqueous sodium hydrogen

carbonate. The organic phase was dried and made up to 100 ml with diethyl ether. Lithium aluminium deuteride (500 mg) was added and the mixture was heated under reflux for 2.5 h. Ethyl acetate and water were then added. The ether was removed under reduced pressure, the residue was taken up in ethyl acetate, washed with dilute hydrochloric acid, water, and aqueous sodium hydrogen carbonate. The solvent was evaporated and the residue dissolved in pyridine (9 ml) and treated with acetic anhydride (4 ml) for 24 h. The solution was poured into dilute hydrochloric acid and the steroids were extracted with ethyl acetate. The extract was washed with water and aqueous sodium hydrogen carbonate and dried. The solvent was evaporated and the residue was chromatographed on silica. Elution with 10% ethyl acetate–light petroleum gave [4 β ,17 α - $^2\text{H}_2$]-3 β ,17 β -diacetoxyandrost-5-ene (710 mg) which crystallized from ethyl acetate–light petroleum as plates, m.p. $158\text{--}160^{\circ}\text{C}$, $[\alpha]_{\text{D}} - 62^{\circ}$ (c 0.4) (lit.,¹² $165\text{--}166^{\circ}\text{C}$, $[\alpha]_{\text{D}} - 69^{\circ}$); δ_{H} (360 MHz) 0.79 (3 H, s, 18-H), 1.02 (3 H, s, 19-H), 2.02 (3 H, s, OAc), 2.03 (3 H, s, OAc), 4.60 (1 H, m, $w_{\frac{1}{2}}$ 15 Hz, 3-H), and 5.36 (1 H, m, 6-H); δ (^2H) 2.29 (4 β -H), 4.59 (17 α -H).

Reaction of [4 β ,17 α - $^2\text{H}_2$]-3 β ,17 β -Diacetoxyandrost-5-ene (13) with Bromine and Silver Acetate.—Bromine (0.1 ml) was added to a solution of [4 β ,17 α - $^2\text{H}_2$]-3 β ,17 β -diacetoxyandrost-5-ene (380 mg) in chloroform (10 ml) at -60°C . Silver acetate (1 g) in pyridine (2 ml) was added in two portions. The mixture was allowed to warm to room temperature and stirred for 24 h in the dark. The mixture was poured into dilute hydrochloric acid and the steroids were extracted with chloroform. The extract was washed with water and aqueous sodium hydrogen carbonate, dried and the solvent was evaporated. The residue was chromatographed on silica. Elution with 25% ethyl acetate–light petroleum gave [17 α - ^2H]-4 β ,17 β -diacetoxy-3 β -hydroxyandrost-5-ene (160 mg) which crystallized from acetone–light petroleum as needles, m.p. $125\text{--}128^{\circ}\text{C}$, $[\alpha]_{\text{D}} - 119^{\circ}$ (c 0.3) (Found: C, 70.8; H, 8.8. $\text{C}_{23}\text{H}_{34}\text{O}_5$ requires C, 70.7; H, 8.8%); v_{max} 3 530, 3 490, 3 380, 1 740, and 1 700 cm^{-1} ; δ_{H} (360 MHz) 0.79 (3 H, s, 18-H), 1.09 (3 H, s, 19-H), 2.03 (3 H, s, OAc), 2.07 (3 H, s, OAc), 3.64 (1 H, m, $w_{\frac{1}{2}}$ 15 Hz, 3-H) 5.37 (1 H, d, J 4 Hz, 4-H), and 5.82 (1 H, m, 6-H); δ (^2H) 4.58 (17- ^2H).

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